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Electronegativity Differences Type of Bond 9. H-O 10. C-H 11. K-F 12. N-H 13. Na-F 14. O-Cl 7-2 Review and Reinforcement (continued) Write Lewis structures for each of the following molecules. Indicated the bonds with either dots or dashes. 15. NH<sub>3</sub> 18. PCl<sub>3</sub>. H-N-H Cl-P-Cl. H Cl 16. H<sub>2</sub> 19. CCl<sub>4</sub>

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review the impact of positive reinforcement on the performances of employees in organizations. It can be applied ... [12]. 4. Intrinsic Reward Another type of positive reinforcement is intrinsic reward. It involves something other than financial motivation. It

The impact of Positive Reinforcement on Employees ...  
14-4 Review and Reinforcement, Class Notes. Terms in this set (11)  
sublimation. conversion of a solid directly into a gas. condensation.  
opposite of vaporization, gas to liquid, exothermic. freezing point.  
temperature at which solid and liquid forms of a substance exist in  
equilibrium.

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4 2 Review and Reinforcement (continued) 12. 13. Could you use a highly intense beam of red light to eject electrons from the surface of a metal? Explain your answer. Give two examples of photoelectric cells and explain how they work. On the line at the left, write the letter of the scientist who made the contribution to the

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Question: 12-3 Review And Reinforcement (continued) Solve The Following Problems In The Space Provided. Show All Your Work  
7.-The Combustion Of Propene Proceeds In Two Steps:  $\Delta H = -124 \text{ kJ mol}^{-1}$  for  $\text{CH}_2=\text{CHCH}_3(\text{g}) + \text{H}_2(\text{g}) \rightarrow \text{CH}_3\text{CH}_2\text{CH}_3(\text{g})$   $\Delta H = -2220 \text{ kJ mol}^{-1}$  for  $\text{CH}_3\text{CH}_2\text{CH}_3(\text{g}) + 5\text{O}_2(\text{g}) \rightarrow 3\text{CO}_2(\text{g}) + 4\text{H}_2\text{O}(\text{l})$  Calculate The Value Of  $\Delta H$  For The Combustion Of 2.70 Mol Of Propene Into Carbon Dioxide And Water Co 224

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