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(d) Calculate the number of moles of precipitate that is produced in the experiment. $1.698 \text{ g} - 1.462 \text{ g} = 0.236 \text{ g PbI}_2$

(s) $2.422 \text{ g PbI}_2 \times \frac{1 \text{ mol PbI}_2}{512.1 \text{ g PbI}_2} = 0.00473 \text{ mol PbI}_2$

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score) and 7 free-response questions, for which you will have 1 hour and 45 minutes (this counts for 50% of your score).

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Exam questions assess the course concepts and skills outlined in the course framework.

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The 2014 AP Chemistry exam was the first administration of a redesigned test as a result of a redesigning of the AP Chemistry course. The exam format is now different from the previous years, with 60 multiple choice questions (now with only four answer choices per question), 3 long free response

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questions, and 4 short free response questions. The new exam has a focus on longer, more in depth, lab-based questions.

~~AP Chemistry - Wikipedia~~

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