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SECTION 1. SHORT ANSWER Answer the following questions in the space provided. 1. ____ The coefficients in a chemical equation represent the.

(a) masses in grams of all reactants and products.

(b) relative number of moles of reactants and products.

(c) number of atoms of each element in each compound in a ...

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Modern Chemistry 77 Stoichiometry CHAPTER 9

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REVIEW Stoichiometry SECTION 3 PROBLEMS Write the answer on the line to the left. Show all your work in the space provided. 1. _____ The actual yield of a reaction is 22 g and the theoretical yield is 25 g. Calculate the percentage yield. 2. 6.0 mol of N₂ are mixed with 12.0 mol of H₂

CHAPTER 9 REVIEW Stoichiometry

CHAPTER 9 REVIEW Stoichiometry SECTION 3

PROBLEMS Write the answer on the line to the left.

Show all your work in the space provided. 1. 88% The actual yield of a reaction is 22 g and the theoretical yield is 25 g. Calculate the percentage yield. 2. 6.0 mol of N₂ are mixed with 12.0 mol of H₂ according to the following equation: N₂(g) + 3H₂(g) ...

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CHAPTER 9 REVIEW stoichiometry (which you studied in Chapter 3) deals with the mass relationships of elements in compounds. Reaction stoichiometry involves the mass relationships between reactants and products in a chemical reaction.

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Modern Chemistry Chapter 9 Stoichiometry Sections 1-3 Introduction to Stoichiometry Ideal Stoichiometric Calculations Limiting Reactant and Percent Yield 2 Definitions. Composition stoichiometry deals with the mass relationships of elements in compounds. Reaction

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Chapter 9 - Modern Chemistry: Stoichiometry -
VOCABULARY** Composition stoichiometry. Reaction
stoichiometry. Mole ratio. Limiting reactant.
calculations involving the mass relationships of
elements in c... calculations involving the mass
relationships between reactant... a conversion factor
that relates the amounts in moles of any t... the
substance that controls the quantity of product that ...

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and ...

Modern Chemistry Chapter 9 Review Stoichiometry
Section 2... Reaction stoichiometry uses molar
relationships to determine the amounts of unknown
reactants or products from the amounts of known
reactants or products.

Chapter 9 Review Stoichiometry Section 2 Answers
Holt Modern Chemistry Review CHAPTER 9:
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(but not all) of the information for the chapter 9 test.
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