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and Equations - Lesson 05 | Balancing Chemical Equations Part 2

Chemical Equations Reactions Section 2

1. They could draw a picture of the chemical reaction. 2. They could write a word equation for the chemical reaction: "Two molecules of hydrogen gas react with one molecule of oxygen gas to

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Section 2 Answers
produce two molecules of water vapor." 3.
They could write the equation in chemical shorthand.
$$2 \text{H}_2 \text{ (g)} + \text{O}_2 \text{ (g)} \rightarrow 2 \text{H}_2\text{O} \text{ (g)}$$

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LibreTexts Section 2 Answers

Chemical Equations and Reactions

SECTION 2 SHORT ANSWER Answer

the following questions in the space

provided. 1. Match the equation type on

the left to its representation on the right. c

synthesis (a) $AX + BY \rightarrow AY + BX$ d

decomposition (b) $A + BX \rightarrow AX + B$ b single-

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displacement (c) $A + B \rightarrow AX + B$ a double-displacement (d) $AX + AY \rightarrow AX + AY$

8 Chemical Equations and Reactions

Here is the balanced symbol equation:

$2\text{Cu} + \text{O}_2 \rightarrow 2\text{CuO}$. You can see that we now have two copper atoms and two

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oxygen atoms on each side. This matches what happens in the reaction:

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continued HOW TO BALANCE A CHEMICAL EQUATION To balance an equation, you must use coefficients. A coefficient is a number that is placed in front of a chemical formula. For example, 2CO represents two carbon monoxide molecules.

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2 SECTION 2 Chemical Formulas and
Equations

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Chemical Formulas and Equations
continued CHECKING SYMBOLS

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When you write a chemical formula, it is important that you check to make sure that it is correct. If you use the wrong formula or symbol in an equation, the equation will not describe the correct reaction. Even a small mis-

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14 SECTION 2 Chemical Formulas and Equations

This is second lecture on a CBSE Class 10th topic - Chemical Reactions and Equations. This chapter is divided into 3 lectures, so make sure you watch the oth...

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Chemistry: Chemical Reactions and Equations (Part 2 ...

the general equation for a single displacement reaction is displacement of the metal in a compound by another metal, displacement of the halogen in a compound by another halogen, displacement of the hydrogen in water by

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Section 2: Chemical equations. A metal species, the displacement of a hydrogen in acid by a metal species

chapter 8 chemical equations and reactions section 2 ...

Section 2: Chemical equations. STUDY. Flashcards. Learn. Write. Spell. Test.

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LaurenMiley2000. Terms in this set (13)
chemical equation. are shorter, easier, way
to show chemical reactions, using symbols
instead of work. What's needed to write an
equation - chemical formulas, reactants,
and products. chemical ...

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4.1: Chemical Reactions and Chemical Equations Last updated; Save as PDF

Page ID 24181; No headers $[2H_2 + O_2 \rightarrow 2H_2O]$ Chemical

formulas and other symbols are used to

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Section 2 Answers
indicate the starting materials, or reactants, which by convention are written on the left side of the equation, and the final compounds, or products, which are written on the right.

4.1: Chemical Reactions and Chemical

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Equations - Chemistry...

Types of reaction Atoms are rearranged during chemical reactions, and are not lost or gained. Chemical reactions can be represented using equations. Catalysts speed up reactions without being used...

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Chemical reactions - Types of reaction -
KS3 Chemistry ...

Ch. 8 Chemical Equations And Reactions

1. Chemical Equations and Reactions

CHAPTER 8 2. What is a chemical

equation? 3. List and describe some

indications of an occurrence of a chemical

reaction. 4. List and describe some

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indications of an occurrence of a chemical reaction.

Ch. 8 Chemical Equations And Reactions
CHEMICAL EQUATIONS AND
REACTIONS 261 SECTION 1 O
BJECTIVES List three observations that

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Suggest that a chemical reaction has taken place. List three requirements for a correctly written chemical equation. Write a word equation and a formula equation for a given chemical reaction. Balance a formula equation by inspection.

Describing Chemical ...

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CHAPTER 8 Chemical Equations and Reactions

Chemical Equations and Reactions

SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. Match the equation type on the left to its representation on the right. a

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b synthesis decomposition single-
displacement double-displacement (a) $AX + BY \rightarrow AY + BX$ (b) $A + BX \rightarrow AX + B$

Home - Kenilworth Public Schools
Chemical reactions are represented by
balanced chemical equations, with

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Chemical formulas symbolizing reactants and products while chemical equations give information in two major areas: first, they tell us what substances are reacting (those being used up) and what substances are products (those being made).

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Section 2 Answers and Chemical Equations - Owlcation ...

Section 2 Chemical Formulas and Equations Key Concept Chemical formulas and chemical equations are used to show how atoms are rearranged to form new substances in a chemical reaction. • Chemical formulas are a simple way to

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Section 2 Answers describe which elements are in a chemical substance. • Chemical equations are a concise way to write how atoms are

Section 2 Chemical Formulas and Equations

The five general types of reaction are

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Solution 2 Answers

combination, decomposition, single-replacement, double-replacement, and combustion. Not all chemical reactions fit uniquely into only one category.

Occasionally, a reaction may fit equally well into two categories. Nevertheless, recognizing a reaction as a particular type is useful.

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11.2 Types of Chemical Reactions - Henry County School ...

It can be balanced by putting a 2 before O₂, giving the balanced equation: CH₄ + 2 O₂ → CO₂ + 2 H₂O. This equation does not have any coefficients in front of

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CH₄ and CO₂, since a coefficient of 1 is dropped. Matrix method. Generally, any chemical equation involving J different molecules can be written as:

Chemical equation - Wikipedia

- Finish Predicting Reactions- worksheet

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#1 (* ANSWERS) *Extra Practice:

Combustion Read 3.3 (p. 137 - 141),

ANSWER p. 141 #41 - 50: Wed, Oct 9:

Continue work from yesterday: Predicting Reactions * Worksheet #2 - Predicting the Products of Chemical Reactions

SUMMARY- Types of Decomposition Reactions (*HANDOUT)

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