

## Chemistry Semester 2 Study Guide

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Chemistry Semester 2 Study Guide. STUDY. PLAY. Kinetic Molecular Theory. 1) gas molecules are small, hard spheres with an insignificant volume 2) the motion of the particles in a gas is random, constant, and rapid 3) all collisions between particles in a gas are perfectly elastic - they are not attracted or repelled my each other

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of most gases is terms of expansion, fluidity, compressibility, and diffusion Chemistry semester

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2 final exam study guide. o Know the difference between directly and inversely proportional. o Know that the pressure and volume of a gas are inversely proportional. o Know how to use Boyle's Law to solve for the initial or final pressure or volume of a.

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Chemistry Semester 2 Final Exam Study Guide ?questionMole answerThe amount of a substance whose mass is equal to the atomic mass in grams questionAvogadro's number. answer6.022 \* 10<sup>23</sup>

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