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In a chemical reaction, one or more

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reactants change into one or more

products. Figure 11.1a shows the ingredients for making leavened bread—flour, salt, yeast, and water. Chemical reactions take place when the ingredients are mixed together and heated in the oven. Figure 11.1b shows the product—a loaf of bread.

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or formulas of the reactants and products  
in a chemical reaction. Writing Chemical  
Equations (pages 321 – 323) 1.

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Chapter 11 Chemical Reactions 113

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reaction occurs when one or more \_\_\_\_\_  
change into one

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complaints are temporary and self-limiting and do not need medication. Moreover, you must cite properties that change as a consequence of the reaction.

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How can she conclude that the reaction started with three nitrogen atoms? There are three nitrogen atoms in the products. The law of conservation of mass states that

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the amount of substance should be the same before and after a reaction.

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### 3 Chemical Reactions and Reaction

Stoichiometry Pp Remember that these answers can be found in different kinds of factors. You may be looking for examples, ratios, weight, sizes, and so on.

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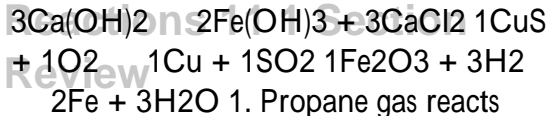
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Problems  $2\text{AgNO}_3 + 1\text{H}_2\text{S} \rightarrow 1\text{Ag}_2\text{S} +$   
 $2\text{HNO}_3$   $3\text{Zn}(\text{OH})_2 + 2\text{H}_3\text{PO}_4$   
 $1\text{Zn}_3(\text{PO}_4)_2 + 6\text{H}_2\text{O} + 2\text{FeCl}_3 +$

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1. Propane gas reacts with oxygen to produce water vapor and carbon dioxide.

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