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for an ionic

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compound represents
one formula unit - the
simplest ratio of the
compound's positive
ions (cations) and its
negative ions
(anions). Monatomic
cations are identified
simply by the
element's name.

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Monatomic anions
more complicated.

The ending of the
element's name is
dropped, the "ide"
added to the root
name. STUDY. PLAY

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formula. subscript of

each element reveals

number of atoms

contained in a single

molecule of a certain

compound.

monoatomic ions.

ions formed from a

single atom;

remember to change

nonmetals to -ide

ending. oxyanions.

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polyatomic ions that
contain oxygen .

binary ionic
compounds.

compounds of 2
elements between
metal ...

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system;... name of the

cation is given first,

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followed by...

polyatomic ions that
contain oxygen;... an
anion that has one
few... monatomic
ions. ions formed
from a single atom.
binary compounds.
compounds
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Describe the contributions made by the following scientists to the development of the periodic table: a.

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b. Dmitri Mendeleev

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c. Henry Moseley. HC

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3 SHORT ANSWER
Answer the following
questions in the space

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provided. 1. Label each of the following statements as True or False: True a. If the formula mass of one molecule is x amu, the molar mass is x g/mol. False b.

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About the Author: DR.

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earned his bachelor ' s from Westmont College, his master of science from California State University, and his EdD from the University of Southern California. He enjoys teaching animal biology, vertebrate biology, wildlife biology,

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