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value of  $q$ , in J, has

the correct number

of significant

figures?  $q = mc \Delta T$

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where  $m = 2.500 \text{ g}$ ,

$c = 4.18 \text{ J g}^{-1} \text{ K}^{-1}$  and

$T = 0.60 \text{ K}$ . A. 6

B. 6.3 C. 6.27 D.

6.270 – 1 Which

molecule has an

index of hydrogen

deficiency (IHD) =

1? A.  $\text{CH}_4$  B.  $\text{CCl}_4$  C.

$\text{C}_2\text{H}_6$  D.  $\text{C}_2\text{H}_2$

$\text{C}_6\text{H}_6$

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19 K 39.10 37 Rb

85.47 55 Cs 132.91

87 Fr (223) Atomic

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number Element

Relative atomic

mass 25 Mn 54.94

43 Tc 98.91 75 Re

186.21 61 Pm

146.92 93 Np (237)

26 Fe 55.85 44 Ru

101.07 76 Os

190.21 62 Sm

150.35 94 Pu (242)

77 I 63 Eu 151.96

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Topic 2 has been

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