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14 Physics in your Kitchen Lab ., I.K. Kikoin, MIR Publishers (1985). 15 How Things Work: The Physics of Everyday Life, Louis A. Bloomfield, John Wiley (2005). 16 Physics Matters: An Introduction to Conceptual Physics, James Trefil and Robert M. Hazen, John Wiley (2004). Bibliography

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where m is the mass of the molecule, T is the absolute temperature of the system, and k is Boltzmann ’ s constant. The probability that the absolute value of the velocity lies between v and v + Δv follows from (1) and has the form. This probability reaches a maximum when . The velocity vo is called the most probable molecular velocity.

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