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Crystallisation. The solution is heated, allowing the solvent to evaporate to leave a saturated solution behind. The saturated solution is allowed to cool slowly and solids will come out of the solution as the solubility decreases, and crystals will grow. Crystals are collected by filtering the solution.

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We spoke to the authors of our new practical workbooks* for Cambridge International AS & A Level Biology, Chemistry and Physics to get their advice and ideas. We asked what practical skills

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There are a few tips you can find in your text book. Other than that, just don't waste your time performing the experiments. Try the simpler looking experiments first (like titration and qualitative analysis). For any experiment that you find a bit complicated, just leave it.

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Halogenoalkanes

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