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Solution From Coulomb's law of electrostatic attraction we know that the PE between two charges Q_1 and Q_2 separated by a distance r is given by $PE = \frac{Q_1 Q_2}{4\pi\epsilon_0 r}$ First we consider the interaction between Na^+ ion and $6Cl^-$ ions at distance r Applying Coulomb's law we have $Q_1 Q_2 = (-6e)(+e) = -6e^2$ $PE_1 = \frac{-6e^2}{4\pi\epsilon_0 r}$

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or $4r$? or $4r$? or Similarly, we now consider 12 Na⁺ ions as second nearest neighbors at a distance r . $Q_1Q_2 (+12e)(+e)$
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Ernest Rutherford (1871-1937) 1.37

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