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1

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Worksheet Solve the

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following stoichiometry

grams-grams problems: 1)

Using the following

equation:  $2 \text{NaOH} + \text{H}_2\text{SO}_4$

$\rightarrow 2 \text{H}_2\text{O} + \text{Na}_2\text{SO}_4$

4 How many grams of sodium

sulfate will be formed if

you start with 200.0 grams

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Worksheet 1 Answers  
of sodium hydroxide and you  
have an excess of sulfuric  
acid?

stoichiometry

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following stoichiometry  
grams-grams problems: 1)

Using the following

equation:  $2 \text{NaOH} + \text{H}_2\text{SO}_4 \rightarrow$

$\text{H}_2\text{O} + \text{Na}_2\text{SO}_4$  How many

grams of sodium sulfate will  
be formed if you start with



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200.0 grams of sodium  
hydroxide and you have an  
excess of sulfuric acid? 2)  
Using the following  
equation:

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Worksheet

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## Stoichiometry Practice

Worksheet Solve the following stoichiometry grams-grams problems: 1) Using the following equation:  $2 \text{NaOH} + \text{H}_2\text{SO}_4 \rightarrow 2 \text{H}_2\text{O} + \text{Na}_2\text{SO}_4$  How many grams of sodium sulfate will

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Worksheet 1 Answers  
be formed if you start with  
200.0 grams of sodium  
hydroxide and you have an  
excess of sulfuric acid? 2)  
Using the following  
equation:

Stoichiometry Practice

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Worksheet With Answers -  
12/2020

Stoichiometry Worksheet and  
Key 1.65 mol  $\text{KClO}_3$  3 mol  $\text{KClO}_3$   
3 mol  $\text{O}_2$  = mol  $\text{O}_2$  2 3.50mol  
 $\text{KCl}$  = mol  $\text{KClO}_3$  3 = 0.275 mol  
 $\text{Fe}$  = mol  $\text{Fe}$  20 3 = = 2  $\text{KClO}_3$   
3 è 2  $\text{KCl}$  + 3  $\text{O}_2$  10. ...

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and key - Saddleback College  
Stoichiometry Practice  
Worksheet Balancing  
Equations and Simple  
Stoichiometry Balance the  
following equations: 1) \_\_\_\_\_

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Worksheet 1 Answers

2)  $\text{N}_2 + \text{F}_2 \rightarrow \text{NF}_3$

3)  $\text{C}_6\text{H}_{10} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$

4)  $\text{Ga}_2\text{O}_3 + \text{NaBr} \rightarrow \text{GaBr}_3 + \text{Na}_2\text{O}$

5)  $\text{SnO} + \text{NF}_3 \rightarrow \text{SnF}_2 + \text{N}_2\text{O}$

6) Solve the following stoichiometry grams-grams problems:

Using the following

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equation:  $2 \text{NaOH} + \text{H}_2\text{SO}_4$   
 $2 \text{H} \dots$

Stoichiometry Practice  
Worksheet

Problem #8: Molten iron and  
carbon monoxide are produced  
in a blast furnace by the

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Worksheet 1 Answers  
reaction of iron(III) oxide and coke (pure carbon). If 25.0 kilograms of pure  $\text{Fe}_2\text{O}_3$  is used, how many kilograms of iron can be produced? The reaction is:  
$$\text{Fe}_2\text{O}_3 + 3\text{C} \rightarrow 2\text{Fe} + 3\text{CO}$$
  
Solution: 1) Determine



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Worksheet 1 Answers:

25000 g / 159.694 g/mol =  
156.5494 mol. 2) Use a ratio  
and proportion to ...

Stoichiometry: Mass-Mass  
Problems #1 - 10  
Mole Conversions and

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## Stoichiometry Review

Worksheet. 1) Using the following equation:  $2 \text{NaOH} + \text{H}_2\text{SO}_4 \rightarrow 2 \text{H}_2\text{O} + \text{Na}_2\text{SO}_4$

4 How many grams of sodium sulfate will be formed if you start with 200 grams of sodium hydroxide and you

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Worksheet 1 Answers

have an excess of sulfuric acid ( $\text{H}_2\text{SO}_4$ )? 2) Using the following equation:  $\text{Pb}(\text{SO}_4)_2 + 4 \text{LiNO}_3 \rightarrow \text{Pb}(\text{NO}_3)_4 + 2 \text{Li} \dots$

Stoichiometry Practice  
Worksheet - Issaquah Connect

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## Stoichiometry Limiting

Reagent Problems #1 - 10.

Limiting Reagent Problems

#11-20 Limiting reagent

tutorial Stoichiometry Menu.

Problem #1: For the

combustion of sucrose: C 12

H 22 O 11 + 12O<sub>2</sub> → 12CO<sub>2</sub>

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Worksheet 1 Answers  
+ 11H<sub>2</sub>O. there are 10.0 g  
of sucrose and 10.0 g of  
oxygen reacting. Which is  
the limiting reagent?

Stoichiometry: Limiting  
Reagent Problems #1 - 10  
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problems name, Mole  
calculation work, Mole mole  
stoichiometry work, Mole  
conversions and



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work and key.

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Please answer the following

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on separate paper using  
proper units and showing all  
work. ... ANSWERS TO  
PROBLEMS Problem 1: a. 0.5 L  
O<sub>2</sub> b. 1.0 L CO<sub>2</sub> Problem 2:  
a. 37.5 L C 2H<sub>2</sub> b. 37.5 L  
H<sub>2</sub>O c. 93.75 L O<sub>2</sub> Problem 3:  
CO<sub>2</sub> = 150 mL, SO<sub>2</sub> = 300 mL

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Worksheet 1 Answers  
Problem 4: a. 0.25 mol H<sub>2</sub>

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Stoichiometry practice  
worksheet. Just what it  
sounds like. How many grams  
of sodium sulfate will be

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formed if you start with 200  
grams of sodium hydroxide  
and you have an excess of  
sulfuric acid. Solution  
stoichiometry worksheet  
solve the following  
solutions stoichiometry  
problems. 1 355 3 grams of

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following balance ...

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chem work 12 2,

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Stoichiometry work 1  
answers, Gas stoichiometry  
work, Stoichiometry work 3.

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answers 1. 0 2 co 2 c.

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Answer the following  
questions on your own paper.

C 4h 10 co 2 e.

Stoichiometry 1 worksheet  
and key. 2 using the  
following equation. Given  
the following equation. C 4h  
10 h 2o 2. O2 h2o d. 2will



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Worksheet 1 Answers  
be formed from 1.65 moles of  
KClO<sub>3</sub>. How many moles of O<sub>2</sub>  
C<sub>4</sub>H<sub>10</sub> CO<sub>2</sub> e. Stoichiometry  
worksheet 1 answers. 2 C<sub>4</sub>H<sub>10</sub>  
13 O<sub>2</sub> 8 CO<sub>2</sub> 10 H<sub>2</sub>O show what  
the following molar ratios  
should be.

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Stoichiometry Worksheet 1  
Answers - Thekidsworksheet  
lesson, they will be more  
likely to identify these  
problems and then solve  
other problems. 14 3 The  
relative strengths of the  
mountain and base -

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stoichiometry section 12.1  
chemistry in the arithmetic  
of equation worksheet  
answers,

source:opentextbc.ca The key  
to remembering here is that  
you need to have some fun  
with this section.

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Chapter 12.1 stoichiometry  
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currently selected item.

Next lesson. Limiting  
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key, Stoichiometry problem  
1, Stoichiometry work 1  
answers, Chapter 6 balancing  
stoich work and key, Chm 130

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Stoichiometry problem 2.

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